

INVITED REVIEW

CELL DAMAGE IN INFLAMMATORY AND INFECTIOUS SITES MIGHT INVOLVE A COORDINATED “CROSS-TALK” AMONG OXIDANTS, MICROBIAL HAEMOLYSINS AND AMIPHILES, CATIONIC PROTEINS, PHOSPHOLIPASES, FATTY ACIDS, PROTEINASES AND CYTOKINES (AN OVERVIEW)

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PROLOGUE

Voluminous literature exists on the mechanisms by which cells and tissues are destroyed in infectious and inflammatory sites. Microbial toxins,^{1–5} their enzymes^{6–9} and cell-wall components,^{10–12} leukocyte and platelets-derived hydrolases and oxygen radicals,^{13–18} cationic polypeptides,^{19–24} arachidonic acid and metabolites, cytokines,^{25,26} coagulation factors and fibrinolysin,²⁷ cytotoxic antibodies and complement components,²⁸ nitric oxide,²⁹ platelet activating factors,³⁰ killer lymphocytes,³¹ as well as additional, still undefined, agonists have all been incriminated as putative agents capable of injuring cells. Special attention has, however, been devoted in the last decade to the role of reactive oxygen species (ROS)¹³ as the main agonists responsible for causing tissue destruction in inflammatory conditions. To date, over 100 human disorders have been connected with an excessive generation of ROS^{13,14,15,32–47}

The pivotal role played by leukocyte-derived ROS, in cellular injury, was often supported by showing that their removal ameliorated and even totally prevented the initiation of cellular damage (see below). The exact nature of the ROS involved in the initiation of cellular damage is, however, still controversial. While superoxide has been advocated as the main toxic oxygen radical,³² H₂O₂, OH·, ROO·, NO and additional more exotic radicals were considered to be the main culprits.^{14–19,33–47} Screening the

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voluminous literature on the role of ROS in cellular injury suggested, however, that these agents, alone, might not be able to reproduce the complex events which contribute to tissue destruction and that a "cross-talk" (see ⁴⁷) among ROS and additional proinflammatory agonists (multiple synergism) might possibly be the main cause of cellular damage in inflammatory and in infectious sites.

A support for this hypothesis stems from studies showing the ability of certain agonists (LPS, cytokines) to "prime" leukocytes to secrete excessive amounts of ROS upon stimulation by additional agonists^{48,49} and from studies on the mechanisms of cellular damage induced by catalase-negative haemolytic streptococci⁵⁰ and by other toxigenic bacteria.^{1,6,8,9}

There are striking similarities between streptococci and activated phagocytes. Both cell types secrete numerous extracellular hydrolases and membrane-damaging agents which allow them to depolymerize the extracellular matrix and the inflammatory exudates and to freely migrate in tissues. Both phagocytes and streptococci possess adhesion molecules, Fc receptors for immunoglobulins and receptors for complement. These facilitate adherence to surfaces of targets and the delivery of toxic agents without the interference of inhibitors present in body fluids.^{39,51} Most importantly, however, both cell types also generate large amounts of H₂O₂ (see ^{14,45,50}).

In 1959 it was demonstrated⁵¹ that H₂O₂-producing group A streptococci which also possessed a potent cell-bound hemolysin (streptolysin S), induced membrane-permeability changes in Ehrlich ascites tumor cells. Such injured cells were disintegrated by the addition of sub-toxic amounts of a streptococcus thiol-dependent proteinase (a synergistic phenomenon). Similarly, streptokinase-activated human plasmin also disintegrated tumor cells which had been injured by complement-dependent cytotoxic antibodies.⁵² These studies suggested that proteinases could amplify cell damage initiated by membrane-damaging agents (a true synergism).

Therefore, the possibility that by analogy to streptococci, activated phagocytes might also injure target cells by employing a well-orchestrated "cross-talk" among their secreted agonists, is plausible.

The purpose of the present review is, to discuss those publications which have described the role played by combinations among ROS and additional putative proinflammatory agonists (microbial hemolysins and amphiphiles, cationic proteins, proteinases, phospholipases, fatty acids, cytokines, xenobiotics) in cellular injury and in the release of membrane-associated lipids.

In Vitro Models of Cellular Injury

Most in vitro models designed to study the mechanisms of cellular damage in infections and inflammation have been conducted with activated neutrophils (PMNs), eosinophiles and with macrophages. A variety of mammalian cells (targets) grown in monolayers were labeled either by ⁵¹chromium (cytotoxicity assay) or with ³H-arachidonic acid (to measure lipid release). Leukocytes were then layered upon the targets to allow adherence and then activated by various agonists (phorbol esters, ionophores, arachidonic acid metabolites, opsonized bacteria, chemotactic peptides, cationic polyelectrolytes, immune complexes). The release of radioactivity from the cells was measured in the supernatant fluids after different periods of incubation. The possible role played by specific leukocyte-derived agonists in cellular injury was evaluated by the inclusion of putative inhibitors (scavengers of ROS, chelators of transition metals, proteinase inhibitors, PLA₂ inhibitors etc). Also, the employment of leukocytes either from patients suffering of chronic granulomatous disease (CGD) or of

myeloperoxidase deficiency (see ¹⁴) yielded important information on the possible role played by selective ROS in cellular injury. It might, therefore be helpful to employ in vitro models, which test the role played by combinations among well-defined agonists representing those released by activated leukocytes and in the presence of appropriate antagonists.

Definitions

Throughout the review, the terms "synergism or synergistic" will be applied to those conditions under which a mixture of two or more of the agonists, tested together, either enhanced cell death or the solubilization of membrane lipids the sum of the values of which was greater than that induced by each of the individual agonists alone. Such coincidental synergy should be differentiated from a true synergism where one toxic agent enhanced the toxic effect of another agent (see ³⁹).

The term "priming"^{48,49} will be used to describe the ability of certain agonists to promote the enhancement of the generation of a certain agonist (e.g. ROS) by leukocytes activated by another agonist (see section K).

Role of Oxidants in Cellular Injury

Voluminous literature has been published on the mechanisms of oxidative cell damage as related to the pathogenesis of inflammatory conditions.^{13 19,32 47} Most publications deal with the role played by ROS generated by activated PMNs, by macrophages and by oxidants generated by the xanthine-xanthine oxidase system.¹⁵ A special attention has also been devoted to the biochemical changes induced in mammalian cells by HOCl^{14,38,39,53,54} and by H₂O₂^{55 68} and to the toxic role played by metal-catalyzed generation of OH· and ROO·.^{15,35,36,67} Oxidants might act directly on membrane lipids to cause oxidation and peroxidation injury,¹⁵ and to enhance membrane fluidity and permeability. Indirectly, H₂O₂ caused a steep drop in adenine nucleotides,^{62,63,65} it inhibited glycolysis^{53 57} it induced mitochondrial swelling⁶⁶ and membrane damage,⁶⁸ and also caused DNA strand breaks (Reviewed in^{55 57}) H₂O₂, also altered signal transduction in mammalian cells.⁶⁸ It activated endothelial cells, selectively by up-regulating intracellular adhesion molecules and major histocompatibility complex class-1,⁶⁹ it modulated the respiratory burst in human PMNs,⁷⁰ it activated complement in normal human serum^{71,72} and phospholipase D in endothelial cells.⁷³ The anti-metabolic effects of H₂O₂ are probably instrumental in lowering the capacity of cells to repair damages. This might expose them to the toxic effects of other leukocyte-derived agonists.

Collaboration Between Microbial Hemolysins and Oxidants

Group A hemolytic streptococci elaborate two potent hemolysins, streptolysin S (SLS)^{1,50,74} and a thiol-dependent and porin-forming streptolysin O (SLO).^{1,50} In addition to their hemolytic activities, both agents also possessed distinct cytopathic properties and also caused disruption of lysosomes (reviewed in⁵⁰). The mechanisms by which SLS and SLO hemolyzed red blood cells and also killed nucleated cells might be due to an alteration of membrane phospholipids,^{1,50} to the elevation of intracellular calcium⁸⁷ and possibly to the activation of PLA₂.^{81,87} The ability of streptococci to produce large amounts of H₂O₂, suggested, therefore, that synergism between hemolysins and peroxide might explain, in part, their distinct virulence.

Cell killing

Ehrlich ascites tumor cells⁵¹ and rat heart cells in culture⁷⁵ which had been exposed either to group A streptococci possessing cell-bound SLS or to cell-free SLS(SLS), developed distinct cytopathic changes. These were characterized by swelling and by the development of large pseudopod-like structures (blebs). On the other hand neither bovine aortic endothelial cells(EC)^{76,77} nor monkey kidney epithelial cells (BGM)^{78,79} were damaged even by large amounts of SLS (50-100 hemolytic units-H.U/ml). However, a distinct synergistic cell killing occurred if either glucose-oxidase,^{76,79} xanthine-xanthine oxidase⁷⁷ or paraquat,⁷⁸ (agents capable of generating H_2O_2), were employed together with SLS. (distinct synergistic reactions). Cytotoxicity was strongly inhibited either by catalase, by trypan blue (an SLS inhibitor) or by phosphatidyl choline.^{80,80} Since EC which had been pretreated with SLS and then washed could still be killed by the addition of H_2O_2 ,⁷⁷ it was assumed that SLS was irreversibly bound to membrane phospholipids,⁸⁰ where it caused subtoxic damage which could be amplified by an oxidant (a synergistic reaction). Unlike SLS, streptolysin O (a thiol-dependent pore-forming toxin) was directly cytotoxic even at a very low concentration (1-2 H.U/ml).⁸¹ However, oxidants further increased cell killing which was blocked by cholesterol. Killing of EC and of BGM cells by mixtures either of SLO or of SLS with H_2O_2 was further increased (also in a synergistic manner) by the addition of trypsin,^{79,81} (a triple synergy, see below), suggesting that oxidants and membrane perforating agents might have affected both proteins and lipids and prepared the grounds for a protease attack.⁸²⁻⁸⁴ The reasons for the relative resistance of both EC and BGM cells to SLS as compared with SLO, is not known.

BGM cells which had been exposed to non-toxic amounts of AAPH (azo bis amidinopropane dihydrochloride) a generator of peroxy radical ($ROO\cdot$)⁶⁷ were also killed in a synergistic manner by the addition of SLS.⁸¹ Cytotoxicity was further enhanced by the addition of trypsin. BGM cells treated simultaneously with GO, AAPH and trypsin were also killed in a synergistic manner.⁸¹ Cell killing by mixtures of SLS, H_2O_2 , and trypsin was inhibited by Mn^{2+} ,⁸⁵ by carrot juice and by a novel PLA_2 inhibitor⁸⁶ suggesting that activation of PLA_2 might be involved in cell killing.⁸⁷

E. Coli hemolysin, also a pore-forming cytotoxin, stimulated human PMNs to generate enhanced amounts of superoxide and also induced the release of elastase and PAF.⁸⁸ The amounts of toxin employed did not hemolyze red blood cells. This hemolysin also proved to be a potent inducer of phosphoinositide hydrolysis. Inositol phosphate and diacylglycerol formation was found to parallel degranulation, the respiratory burst and lipid mediator generation in human PMNs treated by the hemolysin. Its activity was found to be greater than that induced either by f-MLP, PAF, LTB_4 or by the ionophore A23187. The ability of this hemolysin to generate the release of proinflammatory agonists suggested that these agents might also act in concert to enhance cell damage.

Taken together it might be postulated that other toxin-producing microorganisms (staphylococci, clostridia) might also destroy host cells by synergism among their secreted toxins.^{8,9} These findings might perhaps also shed light on the mechanisms by which activated phagocytes induce cellular damage in infectious and in inflammatory sites (see below).

Release of arachidonic acid and metabolites

Mammalian cells treated, even by large amounts of hydrogen peroxide, did not lose

significant amounts of membrane-associated lipids.⁸⁹⁻⁹² The possibility that synergism among oxidants and SLS, which killed cells^{76-79,81} might also result in the solubilization of membrane associated lipids was investigated. Mixtures of SLS and GO-generated H_2O_2 killed both EC and BGM cells but failed to solubilize significant amounts of arachidonic acid unless a proteinase was also present⁷⁹ (a triple synergy). Such mixtures also solubilized large amounts of PGE_2 and 6-keto PGF. The release of arachidonate was strongly inhibited by soybean trypsin inhibitor (SBTI). Although the mechanisms by which trypsin facilitated the release of membrane lipids are not known, it might be speculated that oxidized membrane proteins and lipids might be more readily detached and cleaved by proteinases.⁸²⁻⁸⁴ These results might also suggest that both microbial and leukocyte-derived proteinases might also facilitate the release of large amounts of proinflammatory, arachidonic acid metabolites, even from dead cells which are always present in large numbers in inflammatory sites. This however depends on the additional presence of membrane-damaging agents (hemolysins, phospholipases, free fatty acids, lysophosphatides, cationic proteins).

Collaboration between microbial amphiphiles and oxidants in cell killing

Gram positive and Gram negative bacteria elaborate important surface-associated amphiphiles, lipoteichoic acid (LTA)^{12,93} and lipopolysaccharide (LPS).^{3, 5,94,95} Both amphiphiles can bind spontaneously, via their lipid moieties, to membrane phospholipids^{5,96,97} and to sensitize them to agglutination by antibodies and to a further lysis in the presence of antibodies and complement^{96,98-100} (passive immune kill). Both LTA and LPS served as adhesion agents which facilitated the binding of bacteria to cell surfaces and the delivery of toxic agents. LPS also bound via its lipid moiety to LPS-binding proteins (LBP) present in serum.¹⁰¹ This also enhanced LPS-bearing particles, to associate with the surfaces of monocyte-derived macrophages. Both LTA¹⁰² and LPS^{3, 5,94,95} activated complement and acted as potent stimulators of the generation of tumor necrosis factor.^{4,5,102}

1 Studies with LTA: LTA, which had been released from Gram positive bacteria by phenol, was not cytotoxic to a variety of cells in culture.^{50,96-98} However, endothelial cells (EC) which had been sensitized by LTA and then treated with anti-LTA, IgG, in the absence of complement, could be killed by the addition of subtoxic amounts of GO-generated H_2O_2 .⁷⁹ Paradoxically, however, cell killing also occurred when mixtures of IgG and peroxide were added to EC. Since the toxic activity of the IgG could be removed by adsorption either upon streptococci or upon EC, it was postulated that the IgG preparation might have contained antibodies cross-reactive with EC (Reviewed in⁵⁰).

Human PMNs which had been "primed" by streptococcal LTA, generated synergistic amounts of superoxide, H_2O_2 , and luminol-dependent chemiluminescence and also released large amounts of lysosomal enzymes when treated by anti-LTA globulin.¹⁰³ However, unlike LPS, LTA failed to prime PMNs for enhanced generation of superoxide when challenged either with the chemotactic peptide, fMLP, or with phytohemagglutinin¹⁰³ (see below). Generation of superoxide by LTA-coated PMNs in the presence of anti-LTA IgG was very markedly enhanced by the addition of by-stander, untreated, PMNs. This suggested that migrating PMNs might amplify the respiratory burst by PMNs already coated by LTA.¹⁰⁵ An amplification of superoxide generation by PMN-coated LTA was also observed when small amounts of nuclear histone (a cationic protein) were added together with the anti-LTA, IgG, suggesting

that cationic agents might have "opsonized" the immunoglobulin and facilitated its interaction with the LTA-coated PMN surfaces. Enhanced amounts of superoxide were also generated by PMNs which had been mixed either with LTA-coated fibroblasts or with LTA-coated epithelial cells. This suggested that LTA could be presented to PMNs by non-neutrophilic cells provided that anti-LTA globulin was also present.¹⁰³

LTA derived from a variety of streptococcal and staphylococcal species stimulated peripheral blood monocytes to generate superoxide in the absence of added antibodies.^{104, 106} This suggested that this amphiphile might act as a cross-linker of receptors for LTA present upon the cells. No experiments, however, were performed to test the possible effect of anti-LTA antibodies on superoxide generation by LTA-treated monocytes. The stimulation of monocytes to generate superoxide depended on extracellular calcium and was accompanied by the release of arachidonic acid.¹⁰⁷

Since LTA induced the release of ROS, lysosomal enzymes and a variety of additional agonists, including nitrate^{103, 107} and TNF (102) and since proteinases markedly enhanced the binding of LTA to mammalian cells,¹⁰⁰ it might be speculated that LTA could play an important role in cellular damage caused by Gram positive microorganisms *in vivo* (see⁵⁰).

2 Studies with LPS: The mechanisms by which LPS primed PMNs for enhanced respiratory burst was studied in detail.^{108, 113} The effect of LPS was due neither to an increase in its avidity for NADPH oxidase nor to its affinity for NADPH. The LPS effect was associated with early events of signal transduction, and intracellular Ca^{2+} was a crucial intracellular messenger in LPS priming.¹¹² Since LPS priming was accompanied by a rise in PLA_2 activity which could be inhibited by mepacrine, it was postulated that PLA_2 activation was a fundamental component of priming with LPS. The enhanced generation of superoxide and its dismutation to H_2O_2 following LPS priming might contribute to cellular injury. This might be further enhanced, in a synergistic manner, if additional proinflammatory agonists were also present.

Human platelets incubated with small amounts of LPS released a protein which primed human PMNs for enhanced generation of superoxide following stimulation with f-MLP.¹¹⁰ A short-term exposure of guinea-pig macrophages either to muramyl dipeptide or to LPS caused enhanced release of superoxide in response to immune complexes, but not to PMA.¹⁰⁸ On the other hand, a longer exposure (24 hrs) also resulted in an enhanced superoxide generation by PMA suggesting that the mechanisms of superoxide generation by macrophages stimulated by immune complexes differed from those induced by PMA. Human blood PMNs treated with LPS also showed enhanced expression of C3b receptor, phagocytosis of opsonized bacteria, and peroxide generation, which were blocked by the inhibition of intracellular calcium-dependent processes.¹¹³

Taken together, it might be postulated that amphiphiles (LTA, LPS) release either following exposure of bacteria to leukocyte-derived proteinases^{114, 115} or following bacteriolysis induced either by cationic proteins, phospholipase¹¹⁵ or by penicillin,^{116, 117} might occur *in vivo*. These might contribute to cellular injury via activation of the respiratory burst, the release of lysosomal hydrolases and the activation of the complement cascade. It would also be important to establish whether LTA and LPS which had been released from bacteria, either under physiological or pathological conditions, possessed the same properties as amphiphiles obtained by harsh chemical treatment (e.g. phenol, butanol). Such preparations are traditionally used in experimental models.^{3, 5, 94, 95, 108, 113}

COLLABORATION BETWEEN OXIDANTS AND POLYCATIONS

1 Role of Polycations

Cationic polyelectrolytes are abundantly present in mammalian cells (reviewed and discussed in^{20, 24, 118, 131}). Although most of the polycations possess distinct anti-microbial activities^{22, 24, 118, 121, 124, 125} they were also highly cytotoxic to a variety of both normal and malignant cells.^{127, 131} It is important to note that cationic agents of mammalian cells are always sequestered within membrane-bound structures (azurophilic granules of PMNs, granules of eosinophils, cell nuclei). This prevents them from interacting, by electrostatic forces, with negatively-charged domains present upon all mammalian cells which could cause cell agglutination and damage to their membranes.^{20, 118} In addition to their microbicidal and cytotoxic properties, polycations also function as potent opsonins,^{20, 132, 134} as stimulators of the respiratory burst in leukocytes,^{20, 133, 137} as enhancers of the binding of PMNs to endothelial cells¹³⁸ as potentiating agents in immune-complex cytotoxicity,^{139, 140} as modulators of complement activation¹⁴¹ as enhancers of tissue permeability,¹⁴² as releasers of histamine from basophils,¹⁴³ as stimulators of cytokine generation by a T-cell hybridoma,¹⁴⁴ and as potent stimulators of autolytic wall enzymes in bacteria.¹⁴⁵ In general, polycations were able to agglutinate cells⁹⁴ to induce cross-linking of surface components and to alter membrane permeability by forming ion-channels in lipid bilayers.^{125, 146, 148} The activities of polycations could be inhibited by a variety of polyanions.^{152, 149} The most extensive research on the role of polycations as cytotoxic agents against mammalian cells had been performed with defensins,^{24, 121, 125, 146, 148} with eosinophil cationic proteins,^{120, 122, 123, 141} and to a lesser extent, with synthetic polycations.^{20, 118, 128, 129}

Cytotoxicity induced either by histones^{77, 79} or by defensins^{146, 148} depended on their initial binding, by electrostatic forces, to negatively-charged domains present on the cell surfaces (see¹¹⁸). This could be inhibited either by heparine,¹⁴⁹ by polyglutamate or by polyaspartate. The toxicity of defensins depended on several metabolic events which could be inhibited either by azide, 2-deoxy glucos, cytochalasin B or by lysosomal and calmodulin-mediated inhibitors.¹⁴⁶ This suggested that internalization of the polycation was essential for its cytotoxicity.¹⁴⁶ This, however was unaffected by inhibitors of protein synthesis.¹⁴⁶ On the other hand, inhibitors of endocytosis failed to inhibit the cytotoxicity to endothelial cells induced by synthetic polycations.¹²⁸ Since the cytotoxicity of defensins to targets was strongly inhibited by serum (presumably by albumin),¹⁴⁶ it was postulated that activated PMNs which secreted defensins might injure cells only by an intimate contact, forming intercellular clefts which could exclude inhibitory serum components. (see³⁷).

Paradoxically and unexpectedly, internalization of cationized *Candida albicans* by fibrosarcoma cells which did not elicit a cytotoxic effect, very markedly enhanced tumor growth and spread when injected into mice.¹³³ This enhancement might have been linked with the massive infiltration of neutrophils around the tumors.

The role played by the highly-cationic cathepsin G, elastase, CAP57, Cap 37, lysozyme (cationic antimicrobial proteins) and the slightly cationic, lactoferrin, as microbicidal agents had been recently reviewed^{124, 126} and will not be discussed in the present review.

2 Collaboration Between Polycations and Oxidants

Since activated PMNs and EO generated ROS and also released granule components,

(defensins, leukocyte cationic proteins, MPO, EPO) the possibility that polycations, might collaborate with ROS to enhance cellular damage, is plausible.

Bovine pulmonary endothelial cells (EC), monkey kidney epithelial cells (BGM), pneumocytes, and tumor cells, which had been labeled with ^{51}Cr and then exposed either to a mixture of subtoxic amounts of H_2O_2 , and defensins¹⁵⁰ or to a mixture of peroxide and a variety of cationic agents (histone, polyarginine, lysozyme, ribonuclease, spermine, chlorhexidine, polymyxin B, CETAB), were killed in a synergistic manner.^{77,78} Cell killing could be totally inhibited either by catalase or by polyanions.^{77,78} Similarly, schistosomula of *Schistosoma mansoni* which had been exposed either to oxidant-producing eosinophil cytoplasts or to PMNs were killed in a synergistic manner by the addition of eosinophilic cationic proteins.¹⁵¹ Both the polycations and the oxidant had to be present, simultaneously, to induce synergistic cell killing.^{77,78,150} On the other hand BGM cells which had been exposed to large amounts of histone, to allow binding to membranes and then washed, could still be killed by the addition of H_2O_2 .⁷⁸ However, cells first exposed to glucose oxidase-generated H_2O_2 and then washed, were not killed following the addition of histone.⁷⁷ However, treatment of EC by GO which had been cationized by complexing with poly-L-histidine, and then washed could be killed by the addition of histone⁷⁸ suggesting that the binding of cationized GO to the cell membrane facilitated the direct delivery of peroxide, upon the cell membrane. Cationized GO was also found to be several fold more toxic than unmodified GO, when instilled into the rat trachea.⁸⁵ While serum albumin had a strong inhibitory effect on the killing of targets by mixtures of defensin and peroxide¹⁴⁶ it had only a partial inhibitory effect on the killing of EC induced either by mixtures of histone and peroxide or by mixtures of CETAB and peroxide.⁷⁹ Cytotoxicity induced by mixtures of histone and peroxide was totally inhibited by polyanethole sulfonate.⁷⁷ The toxicity of mixtures of CETAB and peroxide to EC was further enhanced by subtoxic amounts of crystalline trypsin⁷⁹, suggesting that membranes which had been modulated by polycations and by oxidants became more susceptible to proteolysis.^{82,84} It was more difficult to demonstrate⁵⁷ synergism among H_2O_2 , proteinases and histone as the latter was rapidly degraded by the proteinase. This obstacle was overcome by replacing histone by poly-D-lysine which is known to be resistant to trypsin-like enzymes. Under such conditions⁸¹ a distinct synergistic cell killing was also obtained, which was inhibited by soybean trypsin inhibitor (SBTI).

A variety of synthetic polycations (polyarginine, polyhistidine) enhanced the toxicity of immune complexes in a rat model of acute immune complex-mediated tissue damage¹³⁹ (reversed Arthus phenomenon). Similarly to cationized bacteria,^{134,136} the cationized immune complexes also very significantly enhanced the generation of superoxide by PMN, in vitro. Since the infusion of PEG-SOD prior to the induction of the Arthus reaction markedly depressed skin permeability, it was postulated that ROS generated in vivo were involved in the enhanced permeability observed. Also, the cationized immune complexes bound more avidly to the tissues and persisted there for longer periods (see¹³⁶).

The mechanisms by which polycations synergized with oxidants to enhance cellular damage is not fully known. It might, however be postulated that alterations either of the plasma membrane,¹⁴⁸ the induction of cross-linking of surface structures, or the formation of voltage-dependent ion channels (functional "holes"¹⁴⁷) might have allowed the free diffusion of ROS into the cell interior to affect intracellular structures and functions (see^{55,57}).

3 Collaboration between myeloperoxidase and oxidants

Extensive research had been conducted on the toxicity of the highly-cationic neutrophil myeloperoxidase (MPO)^{14,38,39,152-156} and the eosinophil peroxidase (EPO)¹⁵⁷⁻¹⁶² both in vitro and in vivo.

The unique finding that MPO, EPO and lactoperoxidase,¹⁴ when combined with H₂O₂ and with a halide (mostly with Cl and Br) could generate hypohalous acids,¹⁵²⁼¹⁵³ was a major contribution to the understanding of how activated PMNs and eosinophils killed bacteria, parasites but also injured mammalian cells. Both MPO and EPO are highly cationic proteins (PI>9.0). These can avidly bind, by electrostatic forces, to negatively-charged domains present on surfaces of both microorganisms and mammalian cells. While their activities could be inhibited by heme-inhibitors (azide, aminotriazole), the toxicity of HOCl could be inhibited either by taurine or by methionine.³⁹

a) in vitro models: The question of whether HOCl generated by activated PMNs could destroy target cells, via the MPO-H₂O₂-halide system, was investigated.^{39,156,157} While MPO-deficient PMNs were cytotoxic for target cells, PMNs from CGD patients failed to kill cells. However, the cell-free MPO-glucose oxidase system did destroy EC. This was inhibited by scavengers of HOCl suggesting that a prior binding of MPO released from granules might have secured the direct delivery of HOCl upon the targets.^{154,155} It also pointed to the importance of H₂O₂ as a major cytotoxic agent generated by viable PMNs.⁵³⁻⁶³

Either activated leukocytes, cell-free MPO or EPO were cytotoxic to pneumocytes,¹⁵⁸ to tracheal epithelium,¹⁵⁹ and to human nasal epithelium.¹⁶⁰ Cytotoxicity occurred only if the cationic agents had been allowed to bind to the mammalian cells prior to the addition of peroxidase and a halide. Under these conditions toxic oxygen metabolites could be delivered directly upon the targets without the interference of inhibitors.³⁹ Unlike MPO or EPO, other cationic agents (histone, defensins, polycations, cationic detergents) (see above) could kill mammalian targets when simply mixed with peroxide.^{77,79} The damage induced in cells injured by MPO, EPO-H₂O₂ system was characterized by swelling, by the formation of blebs and by exfoliation. Another important observation was that whereas chloride was the most effective halide for MPO killing,³⁹ bromine was more effective than chloride for killing by EPO.¹⁶²

The ability of both native and denatured MPO to function as a cationic agent and as an activator of the autolytic wall enzymes in staphylococci, was demonstrated.¹⁶³ It suggested that high concentrations either of MPO or of EPO might, under certain conditions (e.g. low pH) also function as polycations rather than as an enzymes. It was demonstrated that egg white lysozyme (also a cationic protein) which lost its catalytic property (lysis of *M. lysodeikticus*) still continued to function as a polycation and to activate the autolysis of staphylococci (a charge-dependent reaction).¹⁴⁵

Certain cationic aminoglycoside antibiotics (tobramycin, gentamycin) strongly inhibited the killing of epithelial cells by the MPO, H₂O₂, halide system.¹⁶⁴ Protection against MPO and H₂O₂ was also observed with thioether-containing antibiotics (ticarcillin and teftazimide). The aminoglycosides, which converted HOCl to hydrophilic non-toxic chloramines were unable to prevent the oxidation of sulfhydryls and methionine induced by HOCl. The aminoglycosides protected lung epithelial cells against the cationic MPO by binding to negatively-charged surfaces (a competitive phenomenon).

Treatment of articular cartilage either by reagent H₂O₂ or by peroxide generated by GO, inhibited proteoglycan synthesis but had no effect on its degradation.¹⁶⁵ On the

other hand, HOCl, a product of MPO+H₂O₂+Cl system degraded proteoglycan^{165,166} and also inhibited its synthesis.¹⁶⁵ Cleavage of the proteoglycan core protein was near the hyaluronic acid binding region but there was no evidence that also cleaved glycosaminoglycan chains.¹⁶⁵ Since this effect was partially inhibited by an elastase inhibitor, it was suggested that HClO might act together with proteinases to cause a net increase in degradation (a synergistic reaction).

In vivo models: The role played by the EPO-H₂O₂-halide system in glomerular injury in the rat,^{154,155} in the toxicity to the isolated working rat heart¹⁶² and in bronchoconstriction and vasoconstriction¹⁶⁷ was studied. Tissue alterations depended on the ability of EPO to bind to and to catalyze halogenation of the targets. Since heme inhibitors (azide, aminotriazole) abrogated the pathological changes induced either by MPO or by EPO, it appeared that these highly cationic agents acted not simply as polycations but as catalytic agents which also had the capacity to bind to membranes by electrostatic forces.

4) Mechanisms of Action of HOCl on Mammalian Cells

Since HOCl, the compound generated by MPO, was shown to be a potent toxic agent, the elucidation of its mechanism of action on cells and on the matrix of connective tissue might contribute to the understanding of the pathophysiology of inflammation.^{38,39,60,153-157,165,166,168}

Murine macrophage-like tumor cells in culture which had been treated by 10–20 μM of HOCl showed, 1) oxidation of plasma membrane sulfhydryls, 2) inactivation of glucose and aminoisobutyric acid uptake, 3) loss of cellular K⁺ and 4) and increased cell volume.¹⁶⁸ At higher concentrations (>50 μM), a general oxidation of SH-methionine and of tryptophan residues and the formation of carbonyls was detected. A loss in ATP was also seen. By comparison, NAD degradation and ATP depletion caused by H₂O₂ preceded cell death by several hours. Formation of DNA breaks, a major feature of H₂O₂-induced injury,⁵³⁻⁵⁵ was not observed with HOCl. Thus, targets of HOCl were distinct from those of H₂O₂ with the exception of glyceraldehyde-3-phosphate dehydrogenase which was inactivated by both oxidants.⁵³⁻⁵⁵

The question of whether HOCl generation in physiological environment is detrimental to mammalian cells is controversial. It was suggested³⁹ that despite the absence of tissue toxicity associated with the use of HOCl in the wounded soldiers, this oxidant possessed distinct cytotoxic activity in vitro. At 0.5%, a concentration of HOCl safely used in vivo, a complete in vitro solubilization of skin and liver tissue occurred. However, if this amount of HOCl was added to slices of tissues that had been suspended in wound secretions, the solubilizing action of HOCl was totally lost. In a physiological environment, even millimolar quantities of HOCl preferentially reacted with endogenous amine-containing moieties to yield the derivative chloramine. These newly-generated chloramines exerted strong microbicidal activities but had little, if, any ability to damage normal tissues. The limitation that applied to HOCl-mediated damage in vivo could be circumvented if the highly cationic MPO or EPO could first be bound to their targets where they delivered toxic oxidants, directly upon targets.^{39,153-155} Weiss³⁹ concluded that "given the information, my own enthusiasm for implicating HOCl alone, as the primary key to the neutrophil's tissue destructive potential has been blunted by the surfeit of 70-year-old clinical data". If HOCl can be considered a major toxic oxidant only under strict physiological conditions (a prior binding of MPO), then this oxidant might still function in synergy with other proinflammatory agonists

generated by activated leukocytes. One possibility is that HOCl might inactivate α -1-trypsin inhibitors present in plasma and in the interstitial tissues to facilitate the action of leukocyte-derived elastase, collagenase, gelatinase and cathepsin G on cells and tissues. This might be a true synergism where one agent potentiated the action of another agent. (see ⁶⁹).

COLLABORATION BETWEEN PROTEINASES AND OXIDANTS

PMNs, eosinophils and macrophages secrete a variety of both acid (cathepsin G) and neutral proteinases (elastase, collagenase, gelatinase) which might depolymerize a wide variety of substrates, including extracellular matrix proteins. Proteinases also activated xanthine dehydrogenase to xanthine oxidase,¹⁶⁹ modulated neutrophil-induced superoxide responses,¹⁷⁰ caused cell detachment from matrices^{171 176} and enhanced tumor metastasis.¹⁷⁷ Proteinases might also facilitate hydrolysis of oxidized proteins.^{82 84} The possibility that the simultaneous attack by oxidants, proteinases and by membrane-damaging agents (including xenobiotics) might enhance cellular damage and the release of membrane-associated lipids, is plausible.

a) Role of trypsin: Subtoxic amounts of GO-generated H_2O_2 acted synergistically either with crystalline trypsin, chymotrypsin, cathepsin G or with pancreatic elastase to kill EC in culture.^{178,179} Cell killing was markedly inhibited by ROS scavengers (catalase, dimethylthiourea, Mn^{2+}).⁷⁹ Crystalline trypsin also enhanced, in a synergistic manner, the killing of EC and of BGM cells when combined with oxidants and with a variety of membrane-damaging agents (PLC, PLA_2 lysolecithin),⁷⁹ including ethanol.

b) Role of elastase: Neutrophil-derived elastase might play an important role in the destruction of the matrix of the connective tissue^{39,165,180 187}. This highly cationic enzyme can interact, by electrostatic forces, with negatively-charged domains present in connective tissue matrix. Its activity could be greatly facilitated if both serum α -1-proteinase inhibitor and the secretory leukoproteinase inhibitor,³⁹ were destroyed. Since HOCl generated by the PMN myeloperoxidase- H_2O_2 -halide system could destroy these inhibitors it might be postulated that HOCl generated during PMN activation might be delivered directly upon targets.³⁹ Elastase could also interact with chondrocytes via a specific receptor¹⁸² and endocytosis of elastase (presumably due to its cationic nature) was secondary to cell binding. Studies on the degradation of heparan sulfate proteoglycan of subendothelial matrix, by neutrophil elastase and the MPO- H_2O_2 -chloride system¹⁸⁶ showed that neither elastase alone nor the MPO system alone could induce degradation of proteoglycan. However, when a 4-hour exposure to elastase was followed by a 15 minute exposure to the MPO system, the degradative effect was greater than the additive (a synergistic effect). On the other hand, greater than an additive effect was not observed either when elastase followed the addition of the MPO system or when the two agents were added together. The binding of the cationic elastase to sulfated glycosaminoglycans also resulted in a partial inactivation of the enzyme. It was concluded that PMNs which adhered to glomerular endothelial cells and basement membrane released both ROS and granule enzymes, at the point of attachment. This might have caused damage by a collaboration between oxidants and proteinases.

The question whether leukocyte-derived proteinases also acted in concert with oxidants generated by the NADPH oxidase of phagocytes to destroy matrix, is controversial. The chemotactic peptide fMLP induced the release of MPO from PMNs

which generated HOCl.¹⁸³ The addition of elastase resulted in the release of more matrix components than induced by the individual agents. On the other hand either intact PMNs or cytoplasts which lacked granules but still generated ROS, failed to significantly enhance either elastin or collagen degradation by purified neutrophil elastase. These findings suggested that matrix proteolysis correlated more closely with the release of azurophilic granules than with the generation of ROS by NADPH oxidase.

ROS generated by the xanthine-xanthine oxidase system mediated elastase-induced injury to isolated lung and endothelium.¹⁸⁴ Lungs isolated from rats that had been fed a tungsten-rich diet had negligible levels of xanthine oxidase (XO) activity and after exposure to hyperoxia, developed less acute edematous injury during perfusion with purified elastase. Also, tungsten-treated bovine arterial endothelial cells in culture produced less superoxide and after exposure to neutrophil elastase leaked less albumin than XO-repleted monolayers. It was postulated that tungsten acted by lowering the amount of XO thereby decreasing the amounts of ROS which might have inactivated proteinase inhibitors. This could facilitate the action of elastase (see³⁹).

The mechanisms of PMN damage to human alveolar extracellular matrix (ECM) was studied.¹⁸⁷ Since scavengers or ROS (catalase, SOD) failed to inhibit the solubilization of the ECM, but inhibitors of neutrophil serine protease, elastase and cathepsin G markedly inhibited ECM degradation, it was suggested that ROS might not be involved in matrix degradation.

Lysis of ox RBC by PMA-activated human PMN was reported.¹⁸⁸ RBC lysis correlated with the quantity of a membrane-bound neutral proteinase. Since the solubilized proteinase acted synergistically with added H₂O₂ to lyse the RBC, it was assumed that activated PMNs lysed cells by synergy between oxidants and a cell bound neutral proteinase. This was probably due to a close contact between the RBC and the activated PMN. This is essentially similar to the conditions which allowed the cell bound hemolysins of streptococci to lyse RBC and to kill targets (see above). Also, both MPO and EPO were toxic to targets only if first allowed to bind to the targets.^{154,155}

Sarcoma cells (MCA-1) were killed, in a synergistic manner, when first exposed to non-lytic amounts of H₂O₂ and then treated with a macrophage-derived cytotoxic factor possessing neutral serine proteinase activity.¹⁸⁹ On the other hand, exposure of the cells to the cytotoxic factor prior to the addition of peroxide did not cause cell death. It was suggested that the sequence of the addition of the two factors was important to secure cell death. The necessity to follow a strict order of addition of the reagents might be due to the possibility that a prior exposure to the peroxide diminished the capacity of the targets to cope with a protease attack.

Oxygen metabolites and neutrophil elastase were found to synergistically cause edematous injury in isolated rat lungs.¹⁸⁵ Lung injury could also be shown when PMA-activated PMN cytoplasts (which generated ROS but lacked granules) were infused into the lungs in combination with added PMN elastase.¹⁹¹ It was suggested that elastase was a necessary component for the maximal development of acute edematous injury in isolated lungs perfused with activated PMNs. A similar mechanism might also occur in acute adult respiratory distress syndrome (ARDS). More recently, it was demonstrated that neutrophil elastase acted synergistically with H₂O₂ to inactivate thrombomodulin of endothelial cells. This could contribute to microthrombi formation at the site of inflammation.^{191A} It further demonstrates the important role which might be played by a "cross-talk" among leukocyte-derived agonists in cellular damage in infectious and in inflammatory sites.

The possible role played by proteinases as priming agents for the enhanced generation of superoxide was investigated.¹⁷⁰ Cathepsin G and elastase decreased the order

and increased the lateral mobility of the neutrophil membrane lipids and primed a superoxide response to f-MLP and to PMA. The enhanced amounts of superoxide generated and its dismutation to H_2O_2 might contribute to enhanced cellular damage especially when combined with membrane-damaging agents (see below).

Role of Metalloproteinases

The role played by oxidants in the activation of latent metalloproteinases was investigated.^{39,190-195} This issue was studied in detail by Weiss.³⁹ Metalloproteinases capable of degrading collagen were synthesized in latent, inactive, forms.^{39,195} Unlike PMNs from CGD patients, normal PMNs could activate latent collagenase suggesting that ROS might be responsible for this effect. The active ROS responsible for collagenase activation was identified as HOCl which also activated latent gelatinase. The mechanisms of enzyme activation might involve a thiol-driven intramolecular perturbation that unmasked the active site of the metalloproteinase.¹⁹⁵ Thus the activities of both PMN collagenase and gelatinase were highly linked to the MPO system and HOCl generation (a true synergistic effect³⁹). Although the *in vitro* activation of collagenase by HOCl^{192,193} was corroborated other authors have reported contrary results.^{193a-193b} Furthermore, it was suggested that activation of collagenase was due to cathepsin G and that the role of HOCl was indirect.^{193b} While Weiss *et al.* suggested that gelatinase could also be activated *in vitro* by HOCl,¹⁹² it was recently shown^{193a} that the ratio of HOCl to enzyme employed by Weiss *et al.*¹⁹³ caused distinct inactivation of gelatinase.^{193a} It was recently stressed that in order to activate collagenase under *in vivo* conditions a strict ratio of enzyme/HOCl had to be used^{193a} and that conditions of enzyme inactivation *in vitro* might not be relevant under *in vivo* conditions where HOCl scavengers are present. An endogenous activation of latent collagenase, by plasminogen activator, in rheumatoid synovial cells, was also reported.¹⁹⁶ It suggested that bacteria possessing plasminogen activators (streptococci, staphylococci, clostridia) might also activate metalloproteinases in inflamed sites which might also synergize with oxidants to amplify cell damage. As suggested by Weiss³⁹ "the latent collagenases are specifically structured so that they can transform the oxidized potential of HOCl into the enzyme-catalyzed degradation of the interstitial pericellular, and basement membrane associated collagens."

Collaboration Among Phospholipases, Fatty Acids and Oxidants

PLA₂ and PLC are both implicated in many metabolic and in pathological processes (1,2,6,197-201). The presence of a membrane-associated PLA₂ in PMNs²⁰²⁻²⁰⁴ and the involvement of this enzyme in the respiratory burst which lead to enhanced generation of ROS²⁰⁵⁻²⁰⁷ suggested that it might participate in cellular injury in infectious and in inflammatory conditions. The highly hemolytic phospholipases A₂ and C (PLC) present in venoms of snakes and bees and in certain microbial species, respectively, probably function as distinct cytotoxins and as virulence factors.^{1,2}

Both the free fatty acids and the lysophosphatides which might be released from injured cells following the activation of PLA₂ might injure by-stander cells due to their membrane-damaging properties.

EC and BGM cells which had been pretreated by subtoxic amounts of PLA₂ from pancreas, from snake and bee venoms and by PLC from *Clostridium Welchii*⁷⁶⁻⁷⁹ were killed in a synergistic manner by GO-generated H_2O_2 . Cell killing induced by PLC was further enhanced, also in a synergistic manner, by the addition of crystalline trypsin (a

triple-synergy). Trypsin was also absolutely essential for the release of substantial amounts of tritiated arachidonic acid, PGE_2 and 6-keto PGF from EC treated by mixtures of peroxide and PLC.⁷⁹ Both cell killing and the release of lipids were strongly inhibited by SBTI as well as by catalase, Mn^{2+} and by DMTU.⁷⁹ A similar synergistic cell killing and the release of membrane lipids also occurred when BGM cells were exposed to mixtures of lysophosphatidyl choline (lysolecithin), with peroxide and trypsin.⁷⁹ The possibility that the synergistic killing of BGM cells by mixtures of PLC or SLS with oxidants (H_2O_2 , $\text{ROO}\cdot$) was associated with the activation of the membrane-associated PLA_2 was also investigated. Cells which had been preincubated with a novel PLA_2 inhibitor (carboxymethyl cellulose -CMC, coupled to phosphatidylethanolamine)⁸⁰ became highly refractory to killing by these agents. The PLA_2 inhibitor also depressed the lysis of RBC by SLS. On the other hand, CMC alone, was inactive.

PMNs which had been primed by subtoxic concentrations either of lysophosphatidyl choline (LL), lysophosphatidyl inositol, or of lysophosphatidyl glycerol, generated enhanced amounts of superoxide and H_2O_2 when further stimulated by a variety of agonists (immune complexes, poly-L-histidine fMLP.).^{208,209} These lysophosphatides were also highly hemolytic for human RBC. On the other hand, equi-molar concentrations either of lysophosphatidyl ethanolamine or of lysophosphatidyl serine (phospholipids with polar headgroups) possessed neither hemolytic nor priming activities for enhanced generation of superoxide. However, both these agents when combined with cytochalasin B, became highly stimulatory for superoxide generation.²⁰⁹ Under similar conditions both arachidonic acid and eicosapentanoic acid also possessed stimulatory activities for the enhanced generation of superoxide when tested with PMA.²⁰⁹

The activation of PLA_2 in phagocytes suspended in albumin-containing media might result in the release of free fatty acids, which might synergize with ROS generated by activated phagocytes to kill by-stander cells. Subtoxic amounts either of arachidonic acid, eicosapentanoic acid or of arachidic acid became highly cytotoxic to BGM cells when combined with H_2O_2 and also released substantial amounts of tritiated arachidonated upon the addition of trypsin.⁸¹ Both cell killing and the release of lipids were inhibited by antioxidants as well as by proteinase inhibitors.

Taken together it appears that synergism among phospholipases fatty acids, oxidants and proteinases might lead to cellular damage in inflammatory sites and that lysophosphatides in addition to their direct toxic effect on cells might also serve as priming agents for the enhanced generation of ROS by phagocytes. It might also be speculated that lysophosphatides and free fatty acids released in ischemic and reperfused organs (see^{210,211}) might also act in concert with ROS generated either by the xanthine-xanthine oxidases system or by activated phagocytes, to cause cellular damage.

Collaboration Among Ethanol, Oxidants, Taurocholate and Proteases

The consumption of ethanol was found to be accompanied by the generation of ROS^{44} but the role of the latter in cellular injury induced, by ethanol, in vivo, is still not clear. The possibility that ethanol might synergize with oxidants and with proteinases to kill targets was investigated with BGM and with fibroblasts monolayers.¹¹² Cells which had been exposed to subtoxic amounts of ethanol (6-10%) V/V became rounded but were not killed, even after 4 hours of incubation. On the other hand ethanol at these concentrations synergized with subtoxic amounts either of H_2O_2 or of ROO to rapidly

kill the cells. Cell killing was further enhanced, also in a synergistic manner, by the addition of trypsin (also a triple-synergistic effect). Cell killing was markedly inhibited either by catalase, Mn^{2+} or by fresh carrot juice.

BGM cells which had been treated by mixtures of ethanol, peroxide and trypsin also lost substantial amounts of membrane lipids.⁸¹ These results are essentially similar to those obtained with a series of other membrane-damaging agents (SLS, PLC.LL, PLA₂).⁷⁷⁻⁷⁹

A quadruple-synergism among ethanol, peroxide, taurocholic acid and trypsin also killed BGM cells.⁸¹ A similar mixture of agonists might be found in stomachs of people who consume ethanol and who also have regurgitation of pancreatic juices.^{44,81}

ARE CYTOKINE-PRIMED, PHAGOCYTE-DERIVED OXIDANTS PROTECTIVE OR DESTRUCTIVE?

Since a variety of proinflammatory agonists might act in concert with oxidants to markedly enhance cellular damage, it seems that agents and conditions which enhanced either the synthesis or the secretion of toxic agonists from activated phagocytes or from tissue sources (e.g. xanthine oxidase) might not only better kill microorganisms but might, paradoxically, also destroy the hosts own cells and tissues (*Horror autotoxicus*). Therefore, the flood of investigations on the role played by cytokines as priming agents for the enhanced generation of leukocyte-derived oxygen species,^{25,26,213-250} merit a special consideration regarding the adverse effects that cytokine treatment might cause in patients. Cytokine-primed phagocytes also acquired enhanced phagocytic and degranulation properties,²⁴⁸ higher killing capacities towards a variety of targets²⁵¹⁻²⁵³ and enhanced antibody dependent cellular cytotoxicity properties.²⁵⁴

The biological significance of the cytokine-enhanced generation of oxygen radicals is, however, controversial. While cytokine treatment improved the capacity of hosts to deal with microbial infections, it was also reported that such an elevation was accompanied by complications attributable to enhanced oxidative stress.^{25,26} While some publications showed that chelators of iron were found to inhibit cytokine-enhanced cytotoxicity (involvement of $OH\cdot$) others, have failed to show such an effect. One publication²⁵⁵ showed that the toxicity of TNF was markedly reduced in the presence of H_2O_2 ²⁵⁵ while another showed that superoxide generation by macrophages was depressed by IL-4²⁵⁶ suggesting that this cytokine might regulate some of the effects of the inflammatory response.

Despite the presence of a large body of evidence that cytokine-treated phagocytes generated elevated amounts of ROS and also possessed enhanced cytotoxicity properties, little is known whether cytokine-primed phagocytes might perhaps also generate elevated amounts of other proinflammatory agonists (e.g. cationic proteins, phospholipases, proteinases etc) (see²³⁵). If indeed such an elevation could occur, it might lend support to the assumption that excessive amounts of leukocyte-derived agonists might be deleterious to tissues due to multiple synergistic interactions, as suggested by the models described above.

How Can ROS Generated by Activated Leukocytes Overcome the Antioxidant Systems Present in Targets?

Screening the literature on the mechanisms of cellular damage by activated leukocytes revealed that the amounts of H_2O_2 (micromolar concentrations) generated by activated

PMNs were not sufficient to kill the targets within the regular assay time (2–8 hours). This was especially true if serum proteins were also added to the media. A more significant cell killing occurred after longer periods of incubation (12–18 hrs.) and also necessitated the employment of a high ratio of phagocytes to targets. The high resistance to killing was probably due to the presence of potent antioxidant systems in both targets and in the effector cells.^{257–260} What was commonly observed in in vitro assays, even after overnight incubation, was a detachment of the cell monolayers and the formation of clusters of, mostly viable, targets.^{171–175} The detachment phenomenon was most probably caused by the exocytosis of leukocyte proteinases. This could be totally inhibited by proteinase inhibitors.¹⁷⁴ In order to facilitate cell killing, inhibitors of catalase (sodium azide, amino-triazole¹⁷⁴) and of glutathione reductase¹⁷⁵ had to be added. The removal of proteinase inhibitors by HOCl³⁹ also enhanced cell killing. However, in the absence of such inhibitors, cell killing took a longer time and was probably dependent on synergistic interactions among oxidants, proteinase and additional proinflammatory agonists and perhaps also on a site-specific, radical-induced membrane damage.²⁶¹ The assumption that synergism among oxidants and membrane-damaging agents might be the basis for the understanding how phagocytes kill targets by overcoming their antioxidant capacities, was based on experiments with SLS.⁸¹ While relatively large amounts of H₂O₂ (2–5 mM) failed to kill targets within 2–3 hours, as little as 50 μM of peroxide became cytotoxic within 30 min. if combined with subtoxic amounts of SLS. The mechanisms by which synergism between SLS and peroxide kill cells are still not known. Since SLS could be effectively replaced by other membrane-damaging agents (SLO, fatty acids, lysophosphatides, PLA₂, PLC, cationic proteins, proteinases) (see above) it stands to reason that the synergism phenomenon described is universal and that additional proinflammatory agonists might also be added to the list of agents capable of killing cells.

Morphological Changes Induced by Synergy Among Oxidants Membrane-Damaging Agents and Proteinases

Cellular injury induced by mixtures of oxidants and membrane-damaging agents (lysophosphatides, PLC, SLS, ethanol – see below)^{77–79,212} was characterized by the development of blebs (pseudopod-like structures) usually at one cell pole, which in many cases, were totally detached from the cells. These changes were compatible with altered plasma cell permeability (see 75, 77, 212).

EPILOGUE: WHERE DO WE GO FROM HERE?

Taken together, it might be speculated that the destruction of cells and tissues in infectious and in inflammatory sites might be the result of a well-orchestrated and collaborative events which involve at least three major groups of agonists. 1) oxidants, 2) membrane-damaging agents (hemolysins, phospholipases, lysophosphatides, fatty acids, cationic proteins) and 3) proteinases. All these agents are secreted by bacteria and by activated phagocytes, and their presence in inflammatory exudates, had been demonstrated. The coordinated “cross-talk” among the virulence factors of streptococci, in tissue-destruction, lead to the speculation that invasive and toxigenic bacteria might perhaps be looked upon as “forefathers of modern phagocytes”.^{78,289} H₂O₂^{55–60,262} and HOCl,^{15,39} both long-lasting oxidants, might be the most important toxic agents generated by activated phagocytes. Both oxidants can attack cells from the outside and

also freely diffuse intracellularly to deplete energy sources and to cause DNA breaks.⁵³ Although both $\text{OH}\cdot$ and $\text{ROO}\cdot$ might be more toxic than either peroxide or HOCl ,²⁶⁶ they cannot readily cross the cell membrane unless it had been injured by membrane-damaging agents. Also the effect of $\text{OH}\cdot$ might be dissipated by excess water molecules. Furthermore, its action might also be restricted in a metal-catalyzed site-specific manner.²⁶¹ However, it seems that one of the most important property of H_2O_2 is its ability, at micromolar, sub-toxic concentrations, to synergize with membrane-damaging agents and with proteinases to overcome the antioxidant capacities of targets, to kill cells and to solubilize their membrane lipids.^{77 79,81} Proteinases might act by degrading oxidized proteins more efficiently.^{82 84} H_2O_2 is also capable, at subtoxic amounts, to affect signal transduction in cells⁶⁸ and at concentrations likely to be generated by leukocytes, *in vivo*. Thus the synergism mechanisms proposed might involve amplification of minor damages caused by residual amounts of oxidants which survived the scavenging effects of the cell-associated antioxidant.^{258 460}

Oxidants might perhaps also be involved in programmed cell death (apoptosis),^{269,270} in the oxidation of LDL which become toxic to mammalian cells²⁷¹ and in the killing of targets by NK cells and by other killer lymphocytes. Complement-mediated cell killing might also be amplified by peroxide.⁸¹

To combat the toxic effects of activated leukocytes on tissues, it seems that scavengers of oxidants,^{272 279} might not be adequate as sole therapeutic agents (see²⁷⁶) and that cocktails containing putative inhibitors of membrane-damaging agents and of other "primers" of leukocytes, should be also included in therapeutic regimens. Such agents might either be delivered via regular or cationized liposomes (Ginsburg and Bernholtz unpublished observations) or coupled to cationic polypeptides carriers.^{281 284} Cationization might serve, to increase their adhesiveness and their uptake by targets.²⁰ Finally, it might also be proposed that the ability to effectively combat the deleterious effects of inflammatory agonists *in vivo*, might depend on the ability to disrupt, by mixtures of antagonists, the synergistic "cross-talks" among oxidants, proteinases, phospholipases, fatty acids, cytokines, cationic proteins, microbial toxins and additional membrane-damaging agents, as implied by the experimental models reviewed in the present communication.

Because nitric oxide (NO) has been gaining recognition as an important biological mediator^{29,284,285} and because NO is implicated in the killing of parasites, by activated macrophages (reviewed in²⁸⁵), its role in cellular damage in inflammation should also be considered. NO might interact with superoxide to generate peroxynitrite, a potent cytotoxin, which could degrade to $\text{OH}\cdot$.²⁸⁸ $\text{OH}\cdot$ might then injure targets. Recently, it was demonstrated that BGM cells which had been exposed to subtoxic concentrations of sodium nitroprusside (a generator of NO), were killed, in a synergistic manner, by the addition of mixtures of H_2O_2 and SLS.⁸¹ Hemoglobin, a scavenger of NO, markedly inhibited cell killing. Trypsin further enhanced cell killing, in a synergistic manner, and also markedly enhanced the release of large amounts of membrane lipids. Paradoxically, however, NO was also shown to inactivate superoxide²⁸⁸ suggesting that under certain conditions NO might also function a scavenger of ROS. This paradox should be resolved.

Finally, the findings that mixtures of oxidants and membrane-damaging agents could overcome the antioxidant capacities of mammalian cells, and could also enhance the release of significant amounts of membrane-associated lipids, if proteinases were also included,^{81,289} seem to be important contributions to the understanding the pathophysiology of inflammation. The universal synergy phenomenon described might also be used to devise tissue culture assays for the evaluation of the safety of commonly-used

drugs, pesticides and additional chemicals in common usage. The elucidation of the molecular mechanisms of synergy might also greatly contribute to the formulation of protective therapeutic measures for the purpose of controlling tissue-damaging inflammatory processes.

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